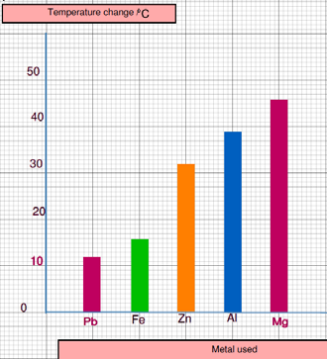


Question	Answers	Notes	Total	Criterion																			
1	a	group 4 period 6		2	A																		
	b	$2PbS + 3O_2 \rightarrow 2PbO + 2SO_2$ Reactants correct Products correct  $2PbO + C \rightarrow 2Pb + CO_2$ Reactants correct Products correct	Do <b>not</b> accept ? in place of a blank coefficient	4	A																		
	c	SO <sub>2</sub> is formed  (which) contributes to the formation of acid rain  <b>or</b>  CO <sub>2</sub> is formed  (which) contributes to climate change	Do <b>not</b> accept "toxic" fumes as this is not specific enough	2	A																		
	d	<b>Any two from,</b> <ul style="list-style-type: none"> <li>• electrical conductivity</li> <li>• thermal conductivity</li> <li>• malleability</li> <li>• appearance</li> </ul>	Any two  Do <b>not</b> accept high melting point for lead	2	A																		
	e	(arsenic is a metalloid/semi-metal so) it would have different structure  different structure gives rise to different properties		2	A																		
	f	<table border="1"> <thead> <tr> <th>Isotope</th> <th>Atomic Number</th> <th>Atomic Mass</th> <th>Protons</th> <th>Electrons</th> <th>Neutrons</th> </tr> </thead> <tbody> <tr> <td><sup>65</sup>As</td> <td>33</td> <td>65</td> <td><b>33</b></td> <td>33</td> <td>32</td> </tr> <tr> <td><sup>70</sup>As</td> <td>33</td> <td>70</td> <td>33</td> <td>33</td> <td><b>37</b></td> </tr> </tbody> </table>	Isotope	Atomic Number	Atomic Mass	Protons	Electrons	Neutrons	<sup>65</sup> As	33	65	<b>33</b>	33	32	<sup>70</sup> As	33	70	33	33	<b>37</b>		2	A
	Isotope	Atomic Number	Atomic Mass	Protons	Electrons	Neutrons																	
<sup>65</sup> As	33	65	<b>33</b>	33	32																		
<sup>70</sup> As	33	70	33	33	<b>37</b>																		
g	Same electron configuration  (So) the different isotopes will react in the same way	WTTE only award the second mark if the first is given	2	A																			

2	a	How does the volatility of esters depend on the length of their hydrocarbon chain?	WTTE – must relate to volatility of esters	1	B
	b	If the length of the carbon chain increases  (then) the volatility of the ester will reduce  (because) the intermolecular forces are stronger	WTTE ECF from part (a), accept a link between chain length and volatility Accept “bonds are stronger”	3	B
	c	<b>Independent variable:</b> ester <b>or</b> length of the carbon chain  <b>Dependent variable:</b> time  <b>Control variables, any three reasonable variables (2 max) for example,</b> <ul style="list-style-type: none"> <li>• temperature</li> <li>• surface area</li> <li>• container size <b>or</b> shape</li> <li>• volume <b>or</b> mass</li> <li>• wind</li> </ul>		4	B
	d	<b>low temperature:</b> no heat added  so avoids breakdown of the ester <b>or</b> position of equilibrium remains towards the right-hand side/shifts right  <b>dry:</b> avoids adding water to the equilibrium mixture /no change in the concentration of water  so avoids breakdown of the ester <b>or</b> so position of equilibrium remains towards the right hand side/shifts right (ORA)	ORA Accept correct answers relating to intermolecular forces for the first two marking points	4	A
	e	<b>A</b>  ethanoic acid  butanol		3	A

3	a	C			1	C	
	b	Thermometer <i>or</i> temperature probe			1	B	
	c		<b>1</b>	<b>2</b>	<b>3</b>	15	B
	<b>Equipment</b>	some equipment is listed	some equipment including a thermometer is listed	appropriate and complete equipment is listed: suitable solutions, thermometer or temperature probe, test tube, measuring cylinder			
	<b>Variables</b>	one control variable is stated	two control variables are stated				
	<b>Method</b>	attempt at a method	temperature is measured	some fine details of technique are included eg recording temperature only when stable, description of how to minimise heat loss			
	<b>Measurements</b>	one metal is investigated	more than two metals are investigated	all five metals are investigated			
<b>Sufficient data</b>	appropriate number of trials	appropriate number of trials and plans to calculate averages					
	<b>Safety</b>	a safety precaution is stated	a safety precaution is stated and linked to hazard				
d	<b>Order:</b> Lead – Iron – Zinc – Aluminium – Magnesium			<i>all correct</i>	1	C	

<b>e</b>	the order of reactivity is based on the temperature rise the larger the <u>temperature change/rise</u> the more reactive the metal is	WTTE	<b>2</b>	C												
<b>f</b>	bar graph		<b>1</b>	C												
<b>g</b>	<table border="1" data-bbox="212 789 586 1024"> <thead> <tr> <th>Metal added</th> <th>Temperature change for the reaction / °C</th> </tr> </thead> <tbody> <tr> <td>Aluminium</td> <td>39</td> </tr> <tr> <td>Iron</td> <td>16</td> </tr> <tr> <td>Lead</td> <td>12</td> </tr> <tr> <td>Magnesium</td> <td>46</td> </tr> <tr> <td>Zinc</td> <td>32</td> </tr> </tbody> </table>  <p>labels for metal on the x axis</p> <p>title linking temperature change to different metals</p> <p>data for one metal correctly plotted</p> <p>data for all metals plotted correctly</p> <p>°C</p>	Metal added	Temperature change for the reaction / °C	Aluminium	39	Iron	16	Lead	12	Magnesium	46	Zinc	32		<b>5</b>	C
Metal added	Temperature change for the reaction / °C															
Aluminium	39															
Iron	16															
Lead	12															
Magnesium	46															
Zinc	32															
<b>h</b>	limited validity as results that produced once only are not reliable as errors may occur more than one trial should be carried out to produce an average	WTTE	<b>2</b>	C												

4	A	test with <u>Lit</u> splint/spill/stick gas burns with a (squeaky) pop (so it is hydrogen)	<i>Accept any reasonable alternative of splint</i>	2	A
	b	if the coin is 100 % copper  (then) it will react the slowest  (because) copper is the least reactive of these metals	<i>Accept "newer coins will react the fastest"</i>	3	B
	c	<b>Independent variable:</b> the composition of the coin <b>or</b> date of coin  <b>Dependent variable:</b> the volume of gas produced in a fixed time <b>or</b> the time taken to produce a fixed volume of gas  <b>Any two reasonable control variables (max 2), for example</b> <ul style="list-style-type: none"> <li>• temperature</li> <li>• concentration of acid</li> <li>• type of acid</li> </ul>	<i>Do not award this mark for volume or time alone</i>  <i>Do not accept pressure</i>	4	B
	d	<b>80s:</b> date 1857-1864  <b>100s:</b> date 1857-1864  <b>70s:</b> date 1962-1982		3	C
	e	<b>Any three reasonable points, for example</b> <ul style="list-style-type: none"> <li>• the coins may not be the exact percentages as indicated in the chart</li> <li>• the results in the calibration graph were for averages and not specific coins</li> <li>• no repeats were possible so there may have been experimental errors</li> <li>• the collection of the gas was inaccurate</li> </ul>		3	C
	f	destructive method  so cannot keep the coin <b>or</b> cannot repeat results	<i>WTTE</i>	2	C

5	a	2.32183 x 10 <sup>4</sup> g	accept 2.32 x 10 <sup>4</sup> , 2.322 x 10 <sup>4</sup> , 2.3218 x 10 <sup>4</sup>	2	D
	b	<p><b>Any three comments about the method, for example</b></p> <ul style="list-style-type: none"> <li>• method is destructive</li> <li>• the method is able to prove the metal is gold</li> <li>• no details of how the solution was measured</li> <li>• no safety precaution given</li> <li>• no repeats possible, method is not reproducible</li> </ul> <p><b>Comment about the validity of the result</b> result is inconclusive <b>or</b> gold could be less than 24 k</p>	WTTE	4	C
	c	<p><b>Any two evaluative points from the list below</b></p> <ul style="list-style-type: none"> <li>• react should be used rather than dissolve <b>and</b> concentrated should be used rather than strong</li> <li>• the hypothesis is invalid because gold is slow to react/unreactive</li> <li>• the relationship between the independent variable / purity of gold / number of Karat and dependent variable (concentration) is correct</li> </ul>	Allow use of concentration or strength for this last marking point	2	C
	d	7.32 g		2	D
	e	7.32/7.89 X100 = 92.8% (to 3 significant figures)		1	C

6	a	<p><i>Grey arrow = fuel + oxygen → carbon dioxide + water</i>  <i>Red arrow = glucose + oxygen → carbon dioxide + water</i>  <i>Green arrow = carbon dioxide + water → glucose + oxygen</i></p> <p>one correct label all labels correct</p>		2	A																					
	b	<p>correct answer 0.44 correct unit °C</p>		2	C																					
	c	<table border="1"> <thead> <tr> <th></th> <th>1</th> <th>2</th> <th>3</th> </tr> </thead> <tbody> <tr> <td><b>Activity that increases the carbon footprint</b></td> <td>An activity that increases the carbon footprint <b>or</b> production of emissions</td> <td>A relevant activity that increases the emissions <b>or</b> increases carbon footprint</td> <td>More than one relevant activity that increases the emission of gases <b>or</b> carbon footprint</td> </tr> <tr> <td><b>Activity that decreases the carbon footprint</b></td> <td>An activity that decreases the carbon footprint</td> <td>A relevant activity that decreases the carbon footprint</td> <td>More than one relevant activity that decreases carbon footprint</td> </tr> <tr> <td><b>Impact on individual/ society</b></td> <td>a general reference to the result of an activity of either an individual <b>or</b> society on the carbon footprint</td> <td>a specific reference to the result of an activity of either an individual <b>or</b> society on the carbon footprint</td> <td>a specific reference to the result of an activity of either an individual <b>and</b> society on the carbon footprint <b>and</b> justification</td> </tr> <tr> <td><b>Explanations</b></td> <td>incomplete scientific explanation</td> <td>complete scientific explanations of impact on carbon footprint for activities that either increase <b>or</b> decrease the carbon footprint</td> <td>complete scientific explanations of impact on carbon footprint for all activities that increase <b>and</b> decrease the carbon footprint</td> </tr> </tbody> </table>				1	2	3	<b>Activity that increases the carbon footprint</b>	An activity that increases the carbon footprint <b>or</b> production of emissions	A relevant activity that increases the emissions <b>or</b> increases carbon footprint	More than one relevant activity that increases the emission of gases <b>or</b> carbon footprint	<b>Activity that decreases the carbon footprint</b>	An activity that decreases the carbon footprint	A relevant activity that decreases the carbon footprint	More than one relevant activity that decreases carbon footprint	<b>Impact on individual/ society</b>	a general reference to the result of an activity of either an individual <b>or</b> society on the carbon footprint	a specific reference to the result of an activity of either an individual <b>or</b> society on the carbon footprint	a specific reference to the result of an activity of either an individual <b>and</b> society on the carbon footprint <b>and</b> justification	<b>Explanations</b>	incomplete scientific explanation	complete scientific explanations of impact on carbon footprint for activities that either increase <b>or</b> decrease the carbon footprint	complete scientific explanations of impact on carbon footprint for all activities that increase <b>and</b> decrease the carbon footprint	12	D
	1	2	3																							
<b>Activity that increases the carbon footprint</b>	An activity that increases the carbon footprint <b>or</b> production of emissions	A relevant activity that increases the emissions <b>or</b> increases carbon footprint	More than one relevant activity that increases the emission of gases <b>or</b> carbon footprint																							
<b>Activity that decreases the carbon footprint</b>	An activity that decreases the carbon footprint	A relevant activity that decreases the carbon footprint	More than one relevant activity that decreases carbon footprint																							
<b>Impact on individual/ society</b>	a general reference to the result of an activity of either an individual <b>or</b> society on the carbon footprint	a specific reference to the result of an activity of either an individual <b>or</b> society on the carbon footprint	a specific reference to the result of an activity of either an individual <b>and</b> society on the carbon footprint <b>and</b> justification																							
<b>Explanations</b>	incomplete scientific explanation	complete scientific explanations of impact on carbon footprint for activities that either increase <b>or</b> decrease the carbon footprint	complete scientific explanations of impact on carbon footprint for all activities that increase <b>and</b> decrease the carbon footprint																							

